

Published in final edited form as:

J Radioanal Nucl Chem. 2016 July; 301(1): 285–291. doi:10.1007/s10967-014-3103-4.

Determination of ²⁴¹Am in Urine Using Sector Field Inductively Coupled Plasma Mass Spectrometry (SF-ICP-MS)

Ge Xiao*, **David Saunders**, **Robert L. Jones**, and **Kathleen L. Caldwell** Inorganic and Radiation Analytical Toxicology Branch, Centers for Disease Control and Prevention, 4770 Buford HWY, Mail Stop F50, Atlanta, GA 30341, USA

Abstract

Quantification of ²⁴¹Am in urine at low levels is important for assessment of individuals' or populations' accidental, environmental, or terrorism-related internal contamination, but no convenient, precise method has been established to rapidly determine these low levels. Here we report a new analytical method to measure ²⁴¹Am as developed and validated at the Centers for Disease Control and Prevention (CDC) by means of the selective retention of Am from urine directly on DGA resin, followed by SF-ICP-MS detection. The method provides rapid results with a Limit of Detection (LOD) of 0.22 pg/L (0.028 Bq/L), which is lower than 1/3 of the C/P CDG for ²⁴¹Am at 5 days post-exposure. The results obtained by this method closely agree with CDC values as measured by Liquid Scintillation Counting, and with National Institute of Standards Technology (NIST) Certified Reference Materials (CRM) target values.

Introduction

Americium is a man-made, radioactive, metallic element produced when plutonium atoms undergo successive neutron capture events in nuclear reactors, in nuclear weapons, and during nuclear weapons' detonations. Americium has several different isotopes, all of which are radioactive. The most important and prevalent americium isotope is ²⁴¹Am, with a half-life of 432.7 years. As it decays, ²⁴¹Am releases alpha particles at 5.44 MeV (13%) and 5.49 MeV (84.5%), becoming ²³⁷Np, which (in 35.9% of the decays) immediately emits gamma radiation at 59.5 keV. The ²⁴¹Am decay chain ends with ²⁰⁹Bi, a nonradioactive element. ²⁴¹Am in the environment originated from atmospheric testing of nuclear weapons during the 1950s and 1960s, as well as reprocessing plants and nuclear accidents. Facilities that are involved with nuclear weapons, well logging sources and manufacture smoke detectors are minor sources of ²⁴¹Am contamination [1].

The critical ²⁴¹Am exposure pathways are inhalation and ingestion. ²⁴¹Am poses significant health hazards, even in small concentrations, if it is taken into the body in a soluble form. Once in the body, ²⁴¹Am concentrates in the skeleton, liver, and muscle. It can stay in the body for decades and continue to expose the surrounding tissues to both alpha and gamma radiation. Long-term internal exposure to ²⁴¹Am may create an increased risk of developing

^{*}Author to whom correspondence should be addressed: gax2@cdc.gov; Fax: + 1 770 488 4097; Tel: + 1 770 488 4100. The Authors declare that they have no competing financial interest.

cancer. Exposure to any significant amount of $^{241}\mathrm{Am}$ is unlikely under normal circumstances [1, 2].

Several techniques exist for the determination of ²⁴¹Am concentration in environmental and human samples [3, 4]. Gamma spectrometry, using High Purity Germanium detectors, is the primary tool used to determine ²⁴¹Am at levels of 0.1–1 Bq/kg or higher; but to obtain accurate results, it requires that the user correct for the attenuation of gamma rays in the samples [5, 6]. Alpha spectrometry is the most commonly applied technique for determination of low-level ²⁴¹Am content. Its principal advantages are relatively low equipment costs, high sensitivity due to low background, and high selectivity for alpha particles against other types of radiation [3]. A limit of detection (LOD) of 10–20 mBq/kg has been reported for various sample matrices, depending on the counting time and count rate of the procedure blank [7–9]. However, tedious, time-consuming sample preparation procedures (e.g., precipitation, evaporation, elution, filtration, electroplating, etc.) and long measurement times limit throughput. Such preparation procedures are due to possible interference from other radionuclides with close alpha energies, and long counting times are necessary because of ²⁴¹Am's relatively low specific activity.

SF-ICP-MS offers substantial advantages over conventional radiometric techniques and has recently been used for analysis of many long-lived radionuclides in various sample matrices. It is one of the fastest methods for ²⁴¹Am analysis [10–13]. An LOD of 1 pg/L for ²⁴¹Am has been reported on SF-ICP-MS [14]. This LOD is comparable to that of alpha spectrometry, assuming no interferences exist for SF-ICP-MS. However, the main analytical issue in SF-ICP-MS originates from isobaric and polyatomic interferences such as ²⁴¹Pu⁺, ²⁴⁰PuH⁺, ²⁰⁹Bi³²S⁺, ²⁰⁹BiO₂⁺, ²⁰⁶Pb³⁵Cl⁺, ²⁰⁴Pb³⁷Cl⁺, ²⁰⁵Tl³⁶Ar⁺, ²⁰⁷Pb³⁴S⁺, and ²⁰¹Hg⁴⁰Ar⁺ [11]. The major isobaric interference with ²⁴¹Am is ²⁴¹Pu. Since ²⁴¹Am is the decay product of ²⁴¹Pu (half-life is 14.33 years), in some samples of reactor origin the concentration of ²⁴¹Am is comparable to that of ²⁴¹Pu. Therefore, the method includes a thorough chemical separation of ²⁴¹Pu (which also removes most of the other interfering molecular ions) from the samples [11, 15].

Developing methods to determine exposure to ²⁴¹Am is within CDC's public health mission. Quantitative analysis of ²⁴¹Am in urine is considered a useful, noninvasive way to assess levels of internal contamination. Our Emergency Response Analytical goal is to be able to detect threat-radionuclides in urine at levels well below (i.e., 1/3 of or lower) the levels for a general population or for special subgroups such as children or pregnant woman (C/P) at the National Council on Radiation Protection & Measurements (NCRP) Report No. 161 Clinical Decision Guide (CDG) based action level of 0.73 pg/L (0.093 Bq/L) for ²⁴¹Am (urine output expected at 5 days post intake) [16]. CDC's IRATB recently developed a urine Gross Alpha/Gross Beta method using Liquid Scintillation Counting (LSC) [17]. However, the LOD of this method for ²⁴¹Am is 4.2 Bq/L, equivalent to 32.3 pg/L, which is much higher than the C/P CDG of 0.73 pg/L for ²⁴¹Am.

In this study, we report a novel and rapid analytical method for determination of ²⁴¹Am in urine samples. The Solid Phase Extraction (SPE) part of the method is based on preliminary studies carried out by Horwitz, et al. [18], Li, et al. [19] and Sadi, et al. [20] using a single

DGA resin cartridge to separate Am from other actinides such as U and Pu. We further optimized the method to isolate ²⁴¹Am from a 10-mL volume of urine using simple extraction steps and used SF-ICP-MS for detection instead of LSC. The study's purpose was to develop a rapid, simple method to address and respond to public health or other accidental, environmental, or terrorism-related exposures to ²⁴¹Am. This method is not designed to characterize the normal background level of ²⁴¹Am in the non-occupationally exposed population, but it does have a detection limit below the suggested CDG action levels. Thus it can serve as a means of rapidly identifying both adults and children who have been exposed to ²⁴¹Am and who might require medical intervention.

Experimental

Reagents and solutions

DGA Cartridges (normal, 1 mL) and a polycarbonate vacuum box (24 holes) were purchased from Eichrom Technologies (Darien, IL, USA). All nitric (HNO₃) and hydrochloric (HCl) acid solutions were prepared from double-distilled (DD) acids (GFS Chemicals Inc. Columbus, OH). Deionized water was used for all solutions (18 MΩ·cm, from an Aqua Solutions Ultrapure Water System, Aqua Solutions, Inc., Jasper, GA). "Base urine" was collected through anonymous human donations (CDC protocol 3994) and acidified to 1% v/v HNO₃. All radioactivity solution sources were traceable to the National Institute for Standards and Technology (NIST, Gaithersburg, MD, USA). Both low and high quality control (QC) solutions and other urine pools for LOD were prepared for determination by spiking base urine with dilutions of an ²⁴¹Am isotope standard (Eckert & Ziegler Analytics, Inc., Atlanta, GA). A series of aqueous ²⁴¹Am Certified Reference Materials (CRM) solutions were prepared by dilution of ²⁴¹Am radioactive source solutions from NIST. ²⁴³Am (Eckert & Ziegler Analytics Inc., Atlanta, GA) was used as an internal standard (tracer). Sodium nitrite (Sigma-Aldrich, St. Louis, MO) was used to adjust the oxidation states. Serial dilutions of uranium, lead, thallium, mercury, bismuth single-element stock standards (SPEX Industries, Inc., Edison, NJ) and a ²⁴²Pu radioactivity solution (U.S. Department of Energy, New Brunswick Laboratory, Argonne, IL) were spiked into the urine samples to verify that high separation factors for U, Pb, Tl, Hg, Bi and Pu were obtained using this SPE procedure.

Sample preparation

The urine sample volume for a single analysis is 10 mL. Allow urine specimens to reach ambient temperature, shake or vortex them to mix for 5 seconds before pipetting. Spike 400 μ L of 1 ng/L ²⁴³Am solution as an internal standard (tracer) to every 10 mL of urine patient sample or QC sample. Add 4.76 mL of concentrated HNO₃ (68–70%, the final concentration in the sample is 5M) and then 0.13g of sodium nitrite to each sample as a valence adjuster to convert Pu to the tetravalent state. Shake or vortex to mix the samples for 5 seconds and let reaction occur at room temperature for at least 10 minutes. Load each sample on a DGA resin cartridge of 1 mL bed volume (cartridge preconditioned with 15 mL of 5 M HNO₃ using a vacuum box). Rinse the cartridge again with 15 mL of 5M HNO₃ followed by 15 mL × 3 of 0.5M HNO₃ using a vacuum box. Strip ²⁴¹Am from the column with 5 mL of 0.5M HCl. Transfer 1 mL of the purified samples into 4 mL polystyrene conical bottom sample

cups for analysis (Figure 1). Prepare external, aqueous-based stock calibration standards by spiking 0.5M HCl with dilutions of ^{241}Am isotope standard, and then add 40 μL of internal standard solution (1 ng/L ^{243}Am) to every 1 mL of standards to reach the same tracer concentration as the patient and QC samples. Prepare both calibration standards and sample blanks as 0.5 M HCl solutions, which match the elute solutions for the column of this method.

Instrumentation

This method measures ²⁴¹Am concentrations using an extended dynamic range, highresolution ICP-MS model Element XR (Thermo Fisher Scientific, Bremen, Germany), which is a double-focusing, magnetic sector, inductively-coupled-plasma mass spectrometer with a single discrete dynode detector (Mascom, Bremen, Germany). It uses the ICP-MS, equipped with nickel sampler and skimmer cones and a CD-2 guard electrode, in triple mode. The sample introduction system consists of a computer-controlled ASX-112 (Cetac, Omaha, NE) autosampler and an Aridus IITM (Cetac, Omaha, NE) desolvation unit. As discussed in our lab's previous report [21], the Aridus IITM setup increases the sensitivity of the SF-ICP-MS by more than 10 times, enabling the measurement of ²⁴¹Am at the low level of < 1 pg/L. Samples self-aspirate from the autosampler into the desolvation unit through an Apex perfluoroalkoxy (PFA) 100 μL/minute nebulizer (ESI, Omaha, NE, or equivalent). The desolvation unit, equipped with an upgraded PFA spray chamber, operates at 110 °C. With the aid of argon sweep gas and nitrogen gas for sensitivity enhancement, the sample passes through a semi-permeable membrane coil in the unit that operates at 160°C. Optimize flow rates as needed, with argon sweep gas at $\sim 3-7$ L/min and nitrogen gas at $\sim 3-7$ mL/min. The desolvated sample exits the unit into a 1.8 mm I.D. sapphire injector and a standard quartz torch, and then into the mass spectrometer. All experimental parameters are optimized for ²⁴¹Am concentrations determination by SF-ICP-MS with respect to maximum ion intensity of ²³⁸U and minimum uranium oxide formation rate using a 5 ng/L natural uranium tuning solution. Table 1 contains a summary of our optimized operating conditions.

Results and discussion

Removal of potential spectral interferences

Potential interferences for analysis of 241 Am include isobaric overlaps with anthropogenic 241 Pu and polyatomic overlaps with 240 PuH+, 209 Bi 32 S+, 209 Bi 02 +, 206 Pb 35 Cl+, 204 Pb 37 Cl+, 205 Tl 36 Ar+, 207 Pb 34 S+, and 201 Hg 40 Ar+. To test for complete removal of 241 Pu, a 50 pg/L solution of 242 Pu isotope spike in base urine was prepared and tested. Experiments showed more than 99% of 242 Pu is removed by the SPE portion of sample preparation. Using SPE sample preparation as described above, Pb, Tl, and Hg, spiked in base urine at concentrations of 3 µg/L, 0.5 µg/L, and 5 µg/L respectively, did not result in apparent (> 0.1 pg/L) 241 Am concentrations. These spiked urine samples' concentrations were above the National Health and Nutrition Examination Survey (NHANES) 95th percentile of urine Pb, Tl, and Hg concentrations [22]. Although no NHANES survey data was available for bismuth, analysis of what was otherwise determined [23, 24] to be a high urine concentration (5 µg/L) of Bi, produced no apparent 241 Am concentration.

Performance of a natural U-spike experiment determined that due to peak tailing, small interferences remained at m/z = 241 when the separated sample solutions contain high levels of U (> 0.5 μ g/L). Analysis of urine samples with U =1.0 μ g/L with this SPE method as part of the sample preparation procedure removed more than 99% of the U that might cause tailing into the m/z=241 region. Samples having U concentrations higher than 10.0 μ g/L (the NHANES 95th percentile of U concentration in urine of normal U. S. residents is 0.031 μ g/L) [22] should be treated by the modified sample preparation procedure as shown in Figure 1, which will be described in more detail below.

Limit of detection

The LOD for ²⁴¹Am in urine specimens is based on 60 analytical runs of 4 different low-concentration samples close to the LOD (a first approximation of LOD is the measured blank concentration plus 3 times the Standard Deviation (SD) of the measured blank concentration) and was calculated according to the formula:

 $Conc_{LOD}$ = [meanb + 1.645(Sb + int)]/[1-1.645(slope)], where mean b = blank average, Sb = standard deviation of blank average, int = intercept of the equation of SD versus concentration for LOD samples analyzed at least 60 times, Slope = slope of the equation of SD versus concentration for LOD samples analyzed at least 60 times.

The LOD of this method is 0.22 pg/L (Figure 2). This LOD is < 1/3 of the C/P CDG (~ 0.734 pg/L), and is therefore acceptable for an emergency radiobioassay method for determining the concentration of 241 Am in urine collected at 5 days post-exposure.

Linearity

A linearity study determined the linear reportable range for this method. The method exhibits good linear signal response between concentrations of 0.3 pg/L and 1000 pg/L of ²⁴¹Am with a Coefficient of Determination of 1.000. The normal calibration range is from 0.3 pg/L to 30 pg/L, and the extended calibration range is from 30 pg/L to 1000 pg/L. If a urine ²⁴¹Am value is above the highest calibrator, the urine sample is diluted with 5% HNO₃ to bring the concentration within the validated calibration range.

Internal methods comparison study

A comparison of urine sample analysis results was performed between this method and our CLIA validated LSC method. The two samples LU-077203 and HU-077201 were prepared as QC material and, using LSC, analyzed for ²⁴¹Am at relatively high concentrations. They then were diluted 1:1000 to get within the desired ²⁴¹Am concentration range for the present method, purified and analyzed using SF-ICP-MS. The difference between the described methods is 2.1% to 3.0% (Table 2),

Precision and accuracy

Analysis of serial aqueous dilutions of a Certified Reference Material (CRM) from NIST was also used to verify method accuracy. The observed ²⁴¹Am concentrations were in close agreement with the target values, with an analytical bias from –0.3% to 1.7% (Table 2). Table 2 also shows the typical precision observed at different concentrations of daily quality

control materials analyzed at the beginning, in the middle, and at the end of each analytical run. Accuracy and precision of the reported results was assured based on adherence to the quality control/quality assurance program of the Division of Laboratory Sciences, NCEH, CDC [25].

Analysis of samples from the NIST Radiochemistry Intercomparison Program (NRIP)

NRIP is a performance evaluation program which provides high quality, traceable radionuclide materials to support low-level radioanalytical laboratories conducting environmental and radiobioassay radioactivity measurements. ²⁴¹Am is among the radionuclides used for testing. However, we found that the extraordinarily high concentrations of uranium present in these samples (intended for evaluation of environmental levels of uranium by alpha spectrometry) significantly affects the accuracy of trace level ²⁴¹Am determination by SF-ICP-MS. Further, these uranium concentrations would possibly produce significant, troublesome instrument contamination. To address this problem we developed and recommend a modified sample preparation procedure that is further optimized for samples with extremely high U content (usually higher than 10 µg/L).

In this procedure, after rinsing the cartridge with 15 mL of 5M HNO $_3$ followed by 15 mL \times 3 of 0.5M HNO $_3$, replace both the cartridge reservoirs and tips to eliminate possible U deposits and rinse the cartridges with more 0.5M HNO $_3$ (15 mL \times (3 - 6) of 0.5M HNO $_3$, see Figure 1). Table 3 and Table 4 show the results observed for 241 Am analysis of the NRIP samples. These samples were from two radiobioassay preparedness exercises during 2012 with different turnaround times (TATs): one 60 days, and one 8 hours. These synthetic urine samples typically have U concentrations ranging from 140 µg/L to 450 µg/L. After the more aggressive rinsing procedure, U concentrations in the elution solutions were under 0.20 µg/L, and did not result in apparent 241 Am signal contribution for these samples. All but one result had slight negative bias (average -2.1 +/-2.4% at a 95% Confidence Level) compared with the NIST target values. Most of the observed results for the NRIP samples show a small negative bias compared to the NIST target values, indicating a slight negative systematic uncertainty. One result had a positive bias of 12.7%. We noted that analyses of this sample for other radionuclides yielded a similar positive bias, indicating an external sample preparation error, as opposed to method bias.

Sample turnaround time (TAT)

While maintaining high quality results, sample TAT is one of the important considerations in a radiological emergency. For this method, ~ 2.5 hours are required to pretreat the urine samples for a batch of 20 patient urine specimens plus QC samples. An additional 3.5 hours are required for final analysis of 20 patient samples by SF-ICP-MS, including calibrators, blanks, and QC samples. Samples may be pretreated concurrently with final SF-ICP-MS analysis, resulting in a daily throughput of approximately 120 samples per day (24 hours) per instrument.

Conclusions

We introduced a method for rapidly determining ultra-low levels of ²⁴¹Am in urine samples using a Solid Phase Extraction purification procedure and a high-sensitivity sample introduction system (Aridus IITM), coupled with SF-ICP-MS. This method provides for analysis of ²⁴¹Am at very low levels, with a LOD of 0.22 pg/L (well below the C/P CDG level) and allows rapid throughput of samples. It attained good agreement, with a bias of 2.1%–3.0%, for urine samples in an internal comparison with a CDC LSC method. It also produced recoveries from 99.7% to 101.7% in analysis of aqueous dilutions of ²⁴¹Am SRM from NIST.

This method's efficient urine sample separation scheme effectively eliminates most molecular ion interferences. However, if urine samples contain more than $10 \,\mu\text{g/L}$ of U, more aggressive rinsing procedures are required to eliminate U from the elution solutions. The results obtained by this method for NIST/NRIP reference materials with high-U levels are in close agreement with the NIST target values, with biases ranging from -0.62% to -5.61%.

A major advantage of this method over alpha spectrometry and other methods is that only a small, 10 mL volume of each urine sample is needed to perform the analysis, making successful analysis more likely, especially for young children and infants.

This procedure is appropriate for rapid identification and quantification of ²⁴¹Am in urine for emergency response involving accidental or terrorism-related elevated exposures, or for evaluating chronic environmental or other non-occupational exposures.

Acknowledgments

The authors thank Baki B. Sadi at the Radiation Protection Bureau, Health Canada and Ted Zateslo at the Thermo Fisher Scientific technical support group (U.S.) for their help and technical assistance.

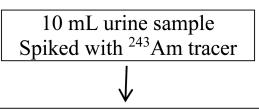
The opinions expressed in this article are the author's own and do not reflect the view of the Centers for Disease Control and Prevention, the Department of Health and Human Services, or the United States government.

References

- 1. [Last Accessed on 9/10/2013] http://www.epa.gov/radiation/radionuclides/americium.html
- 2. [Last Accessed on 9/10/2013] http://www.atsdr.cdc.gov/phs/phs.asp?id=809&tid=158
- 3. Vajda N, Kim C-K. Journal of Radioanalytical and Nuclear Chemistry. 2010; 284:341–366.
- 4. Hou X, Roos P. Analytica Chimica Acta. 2008; 608:105-139. [PubMed: 18215644]
- Appleby PG, Richardson N, Nolan PJ. Nuclear Instruments & Methods in Physics Research Section B-Beam Interactions with Materials and Atoms. 1992; 71:228–233.
- 6. Canberra Genie (2000) software package self-absorption correction.
- Kumar R, Yadav JR, Rao DD. Journal of Radioanalytical and Nuclear Chemistry. 2011; 289:451– 454.
- 8. Dai X, Kramer-Tremblay S. Journal of Radioanalytical and Nuclear Chemistry. 2011; 289:461-466.
- Warwick PE, Croudace IW, Oh JS. Analytical chemistry. 2001; 73:3410–3416. [PubMed: 11476242]
- 10. Varga Z, Suranyi G, Vajda N, Stefanka Z. Microchemical Journal. 2007; 85:39-45.
- 11. Varga Z. Analytica Chimica Acta. 2007; 587:165–169. [PubMed: 17386769]

12. Agarande M, Benzoubir S, Bouisset P, Calmet D. Applied Radiation and Isotopes. 2001; 55:161–165. [PubMed: 11393755]

- 13. Horwitz EP, Chiarizia R, Dietz ML. Reactive & Functional Polymers. 1997; 33:25-36.
- 14. La Rosa JJ, Burnett W, Lee SH, Levy I, Gastaud J, Povinec PP. Journal of Radioanalytical and Nuclear Chemistry. 2001; 248:765–770.
- 15. Pourmand A, Dauphas N. Talanta. 2010; 81:741-753. [PubMed: 20298848]
- Management of Persons Contaminated With Radionuclides: Handbook. 2008. Report No. 161 I, p. 158, ISBN-13: 978-0-929600-99-4, Executive Director: DA Schauer
- 17. Piraner, O. CDC CLIA Urine Gross Alpha-Beta. 2009 Feburary. 3011.1
- 18. Horwitz EP, McAlister DR, Bond AH, Barrans RE. Solvent Extraction and Ion Exchange. 2005; 23:319–344.
- 19. Li C, Sadi B, Benkhedda K, St-Amant N, Moodie G, Ko R, DiNardo A, Kramer G. Radiation Protection Dosimetry. 2010; 141:228–232. [PubMed: 20573683]
- Sadi BB, Li C, Masoud A, Ko R, Kramer GH. Radiation Protection Dosimetry. 2010; 141:134–139. [PubMed: 20488975]
- 21. Xiao G, Jones RL, Saunders D, Caldwell KL. Radiation Protection Dosimetry. 2014
- 22. [Last accessed on 02/25/2014] The Fourth National Report on Human Exposure to Environmental Chemicals. 2009. http://www.cdc.gov/exposurereport/pdf/FourthReport.pdf
- 23. Serfontein WJ, Mekel R, Bank S, Barbezat G, Novis B. Res Commun Chem Pathol Pharmacol. 1979; 26(2):383–389. [PubMed: 523778]
- 24. Carson, BL.; Ellis, HV.; McCann, JL. Toxicology and Biological Monitoring of Metals in Humans. Lewis Publishers, Inc.; 1986. p. 44-47.
- 25. Caudill SP, Schleicher RL, Pirkle JL. Stat Med. 2008; 27(20):4094-4106. [PubMed: 18344178]

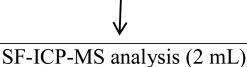


Oxidation with 4.76 mL double distilled HNO₃ (68-70%) / 0.13 g NaNO₂



Matrix removal and separation with DGA resin cartridge

- 1. Resin conditioning with 15 mL of 5M HNO₃
- 2. Loading of the urine sample
- 3. Wash: 15 mL of 5M HNO₃
- 4. Wash: 15 mL x 3 of 0.5 M HNO₃
- 5. Replace: cartridge reservoir and tips*
- 6. Wash: 15 mL x (3-6) of 0.5 M HNO₃*
- 7. Am elution with 5 mL of 0.5 M HCl



Sequential sample preparation procedure for ²⁴¹Am determination * Samples containing U concentrations greater than 10 µg/L, add steps 5 – 6.

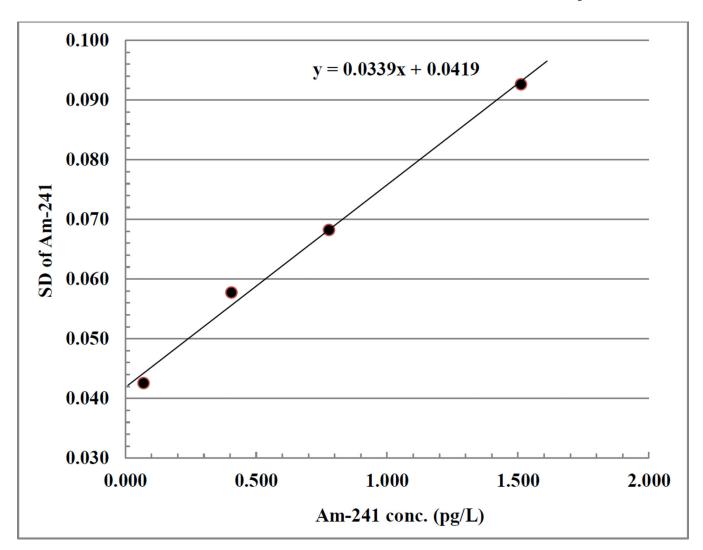


Figure 2. Plot for ²⁴¹Am LOD determination (60 runs per point).

Xiao et al. Page 11

Table 1

Instrumental conditions and data acquisition settings for SF-ICP-MS measurements

| RF Power (KW) | 1.2 – 1.3 | |
|------------------------------|--------------------------------------|--|
| Cooling Gas flow (L/min) | 16 | |
| Auxiliary Gas flow (L/min) | 0.9 | |
| Sample Gas flow (L/min) | 0.7 - 0.8 | |
| Lenses (V) | Optimized as needed | |
| Sample Take up time (min) | 2.1 | |
| Wash (min) | 3 | |
| Pump Speed During Wash (rpm) | 1 | |
| LR Runs/Passes | 3* 60 | |
| Detection Mode | Triple | |
| Measurement Units | CPS | |
| Scan Type | ESCAN | |
| Scan Optimization | Speed | |
| Number of Pre-Scans | 5 | |
| Integration Type | Average | |
| Res. Switch Delay (s) | 2 | |
| Resolution | 300 | |
| Mass Window (%) | 15 | |
| Setting Time (s) | 0.001 | |
| Sample Time (s) | 0.001 | |
| Samples Per Peak | 200 | |
| Search Window (%) | 20 | |
| Integration Window (%) | 15 | |
| Measured Isotopes | ²⁴¹ Am, ²⁴³ Am | |
| | | |

Table 2

Observed ²⁴¹Am concentrations (pg/L) and among-run precision for reference materials and internal quality control materials

| N Average SD 3a 12 8.220 310 1a 12 20,700 600 59 0.335 0.058 c 120 0.708 0.068 c 120 1.44 0.093 c 120 1.44 0.093 f 10 783 33.6 f 1 100 - f 1 300 - f 1 339 - f 1 399 - f 1 605 - f 1 1007 - | į | | | | ²⁴¹ Am | |
|---|---|-----|---------|---------------|--------------------|----------|
| 12 8,220 310 12 20,700 600 59 0.335 0.058 120 0.708 0.068 60 1.44 0.093 120 10.1 0.56 40 783 33.6 1 203 - 1 300 - 1 605 - 1 605 - 1 1007 - 1 1007 - | Sample | Z | Average | \mathbf{SD} | Target Value | Bias (%) |
| 12 20,700 600 59 0.335 0.058 120 0.708 0.068 60 1.44 0.093 120 10.1 0.56 1 100 - 1 203 - 1 300 - 1 605 - 1 605 - 1 1007 - | ${ m LU}$ -077203 a | 12 | 8,220 | 310 | 8,050 ^d | 2.1 |
| 59 0.335 0.058 120 0.708 0.068 60 1.44 0.093 120 10.1 0.56 40 783 33.6 1 100 - 1 203 - 1 300 - 1 605 - 1 1007 - | HU -077201 a | 12 | 20,700 | 009 | $20,100^{d}$ | 3.0 |
| 120 0.708 0.068 60 1.44 0.093 120 10.1 0.56 40 783 33.6 1 100 - 1 203 - 1 399 - 1 605 - 1 1007 - | $\operatorname{Pooll}{b}$ | 59 | 0.335 | 0.058 | 0.3^{e} | 12 |
| 60 1.44 0.093 120 10.1 0.56 40 783 33.6 1 100 - 1 203 - 1 390 - 1 605 - 1 605 - 1 1007 - | Low QC c | 120 | 0.708 | 0.068 | 0.7^{e} | 1.2 |
| 120 10.1 0.56 40 783 33.6 1 100 - 1 203 - 1 300 - 1 399 - 1 605 - 1 1007 - | $\mathrm{Pool2}^{b}$ | 09 | 1.44 | 0.093 | 1.4e | 3.0 |
| 40 783 33.6 1 100 - 1 203 - 1 300 - 1 399 - 1 605 - 1 1007 - | High QC $^{\mathcal{C}}$ | 120 | 10.1 | 0.56 | 10.0^{e} | 6.0 |
| 1 100 – 11 203 – 11 300 – 11 300 – 11 399 – 11 605 – 11 1007 – 11 | Extended High $\mathrm{QC}^\mathcal{C}$ | 40 | 783 | 33.6 | 800^{e} | -2.1 |
| 1 203 - 1 1 300 - 1 1 399 - 1 1 605 - 1 | Dilution 1^f | _ | 100 | ı | 100 | 0.0 |
| 1 300 – 1 399 – 1 605 – 1 1007 – | Dilution 2^f | - | 203 | I | 200 | 1.7 |
| 1 399 – 1 605 – 1 1007 – | Dilution 3^f | _ | 300 | ı | 300 | -0.1 |
| 1 605 - 1 1007 - | Dilution 4^f | _ | 399 | I | 400 | -0.3 |
| 1 1007 – | Dilution 5^f | - | 909 | I | 009 | 0.8 |
| | Dilution 6^f | _ | 1007 | ı | 1000 | 0.7 |

 $^{^{2}\}mathrm{I:}1000$ dilution of urine QC materials used for the LSC Gross Alpha/Beta method at CDC.

 $[^]b$ Urine materials made at CDC by spiking certified reference material in pooled urine collected anonymously.

^CInternal quality control materials made at CDC by spiking certified reference material in pooled urine collected anonymously.

d. Characterized results of co-worker by using the LSC Gross Alpha/Beta method at CDC[17].

 $[\]stackrel{e}{r}$ Target values of spiked urine pools using certified reference material.

fAqueous dilutions of CRM from NIST.

Xiao et al.

Page 13

| Sample ID | Massic Activity (NIST Target Value) | Massic Activity (CDC Observed Results) | Relative Expanded Uncertainty (k=2) | Bias |
|-----------|--|---|--|-------|
| | (Bq/g spike) | (Bq/g spike) | (%) | (%) |
| 207 | 1.784 | 1.74 | 11.5 | -2.52 |
| 212 | 1.784 | 1.74 | 11.2 | -2.41 |
| 220 | 1.784 | 1.76 | 12.8 | -1.57 |
| 224 | 1.784 | 1.74 | 12.5 | -2.52 |
| 227 | 1.784 | 1.76 | 12.1 | -1.63 |
| 214 | 1.784 | 1.74 | 12.3 | -2.47 |
| 216 | 1.784 | 1.71 | 12.1 | -4.09 |
| 228 | 1.784 | 1.75 | 12.2 | -2.02 |
| 231 | 1.784 | 1.77 | 11.5 | -0.95 |
| 232 | 1.784 | 1.77 | 11.2 | -0.73 |
| 208 | 1.784 | 1.74 | 13.1 | -2.75 |
| 211 | 1.784 | 1.74 | 13.0 | -2.58 |
| 219 | 1.784 | 1.68 | 13.8 | -5.61 |
| 223 | 1.784 | 1.76 | 11.7 | -1.57 |
| 226 | 1.784 | 1.77 | 11.8 | -1.01 |

^{*} All samples were diluted 1:2 before DGA (Eichrom's extraction chromatographic materials in which the extractant system is N,N,N',N'-tetra-noctyldiglycolamide resin) separation

Table 4 Comparison of CDC 241 Am results with NIST target values for NRIP12 8 hours samples *

| Sample ID | Massic Activity (NIST Target Values) | Massic Activity (CDC Observed Results) | Relative Expanded Uncertainty (k=2) | Bias |
|-----------|---|---|--|-------|
| | (Bq/sample) | (Bq/sample) | (%) | (%) |
| 215 | 0.146 | 0.14 | 11.0 | -1.93 |
| 218 | 0.292 | 0.29 | 10.9 | -0.62 |
| 222 | 0.149 | 0.15 | 11.1 | -1.47 |
| 230 | 0.297 | 0.29 | 11.1 | -1.99 |
| 234** | 0.372 | 0.42 | 11.2 | 12.7 |

^{*} Samples 218 and 230 were diluted 1:2 and sample 234 was diluted 1:4 before DGA (Eichrom's extraction chromatographic materials in which the extractant system is N,N,N',N'-tetra-n-octyldiglycolamide resin) separation

^{**}Analyses for other radionuclides also produced unusually high results for this sample.